

# Influence of feedstock demineralization on the chemical composition of pyrolysis vapours derived from sugarcane residues in py-GC/MS

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# Lignocellulosic materials obtained from sugarcane processing



Mechanical harvesting (75%)

Sugarcane trash (SCT)

Leaves and cane tops



Extraction of the juice

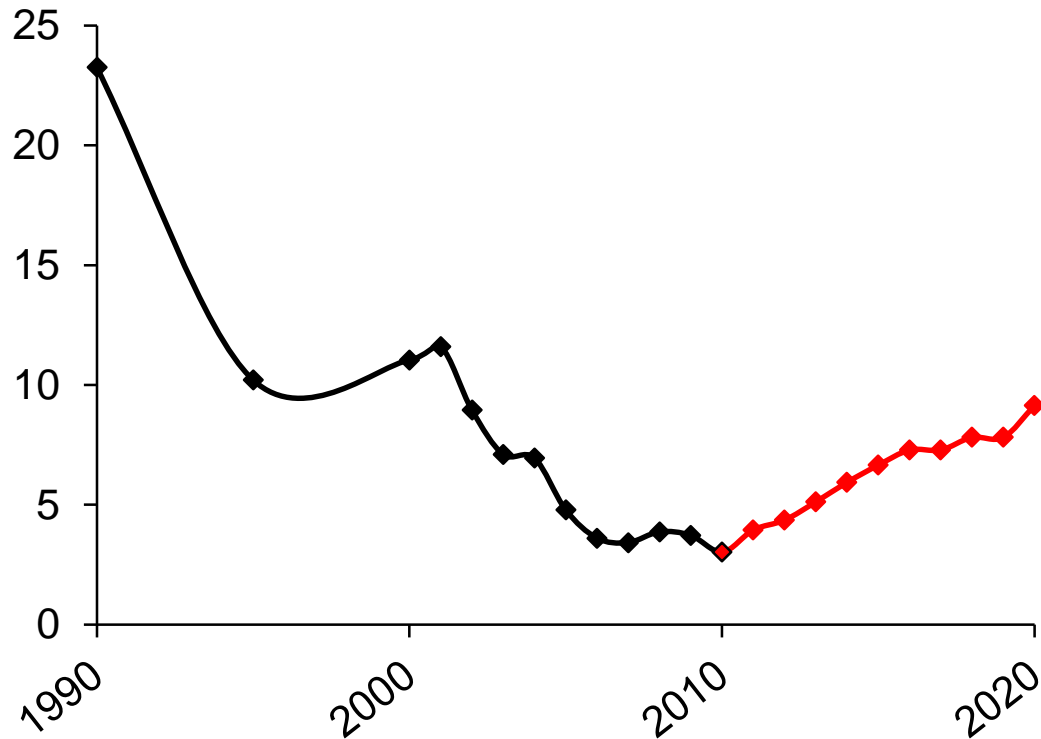


Plant stalk



Sugarcane bagasse (SCB)

# Production of sugarcane bagasse in Cuba (Mton)



Statistical Yearbook of Cuba, 2015

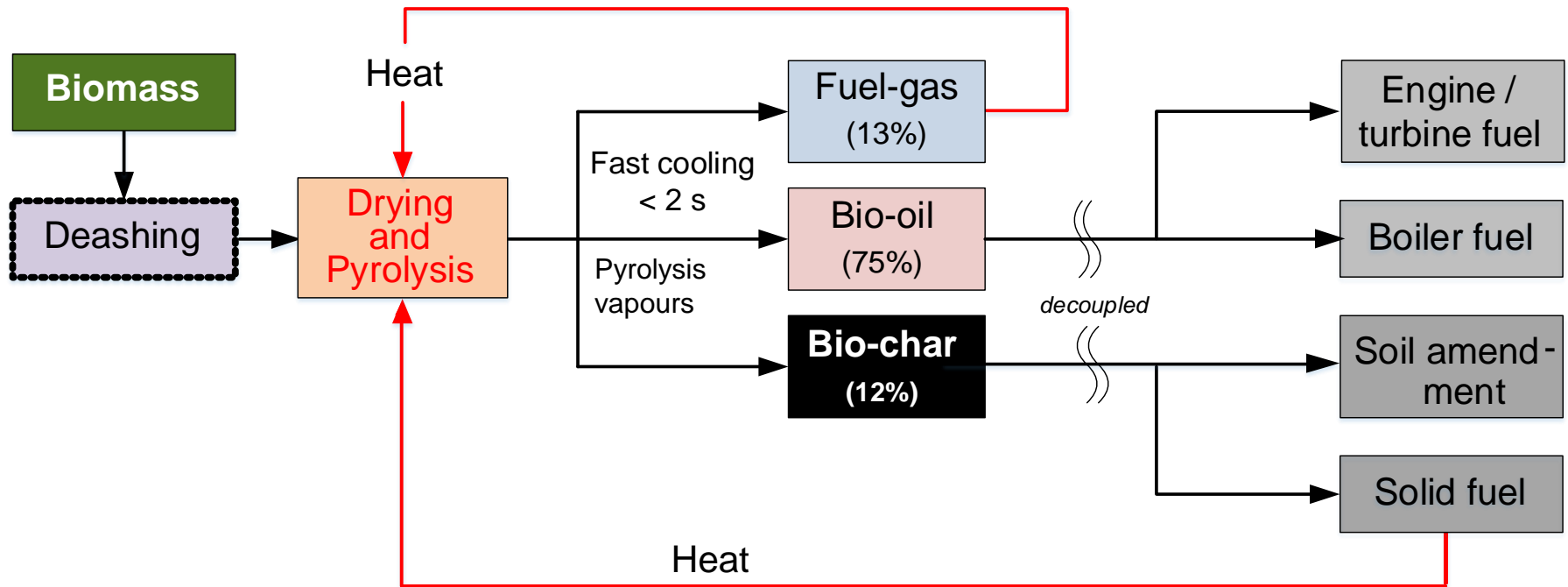


Sugarcane biomass residues	Availability, ton/ton crushed cane (Use)	Surplus available
Bagasse + trash in sugar mill	<b>0.32</b> (Steam supply and power generation)	<b>15-20%</b>
Trash left in field	<b>0.17</b> (Improve soil quality)	<b>?</b>
Trash separated in cane cleaning centers	<b>0.11</b> (Power generation)	<b>40%</b>



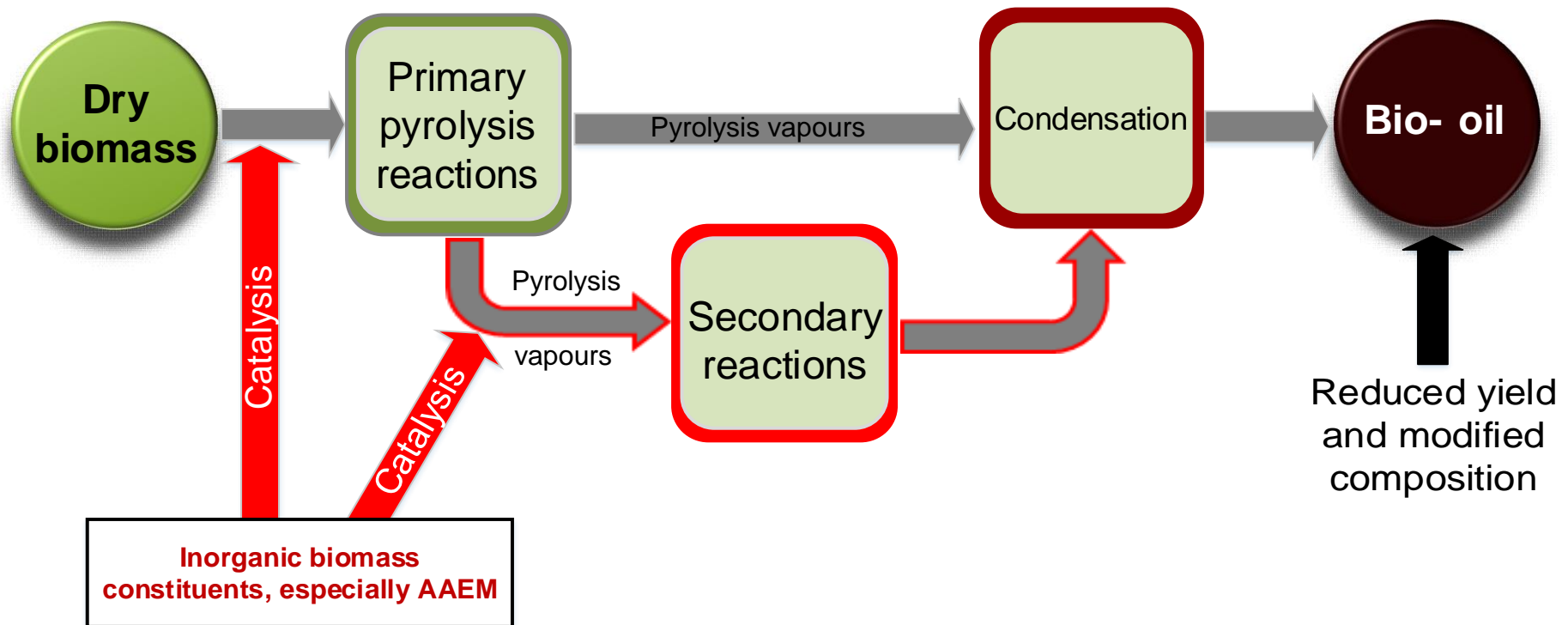


# Fast pyrolysis fast platform for sugarcane lignocellulosic residues



Rodriguez et al., 1987. Sugar cane bagasse as a feedstock for an industrial fast pyrolysis process under development. *Journal of Analytical and Applied Pyrolysis*, 12, 301-318

# Influence of inorganic biomass constituents on pyrolysis results

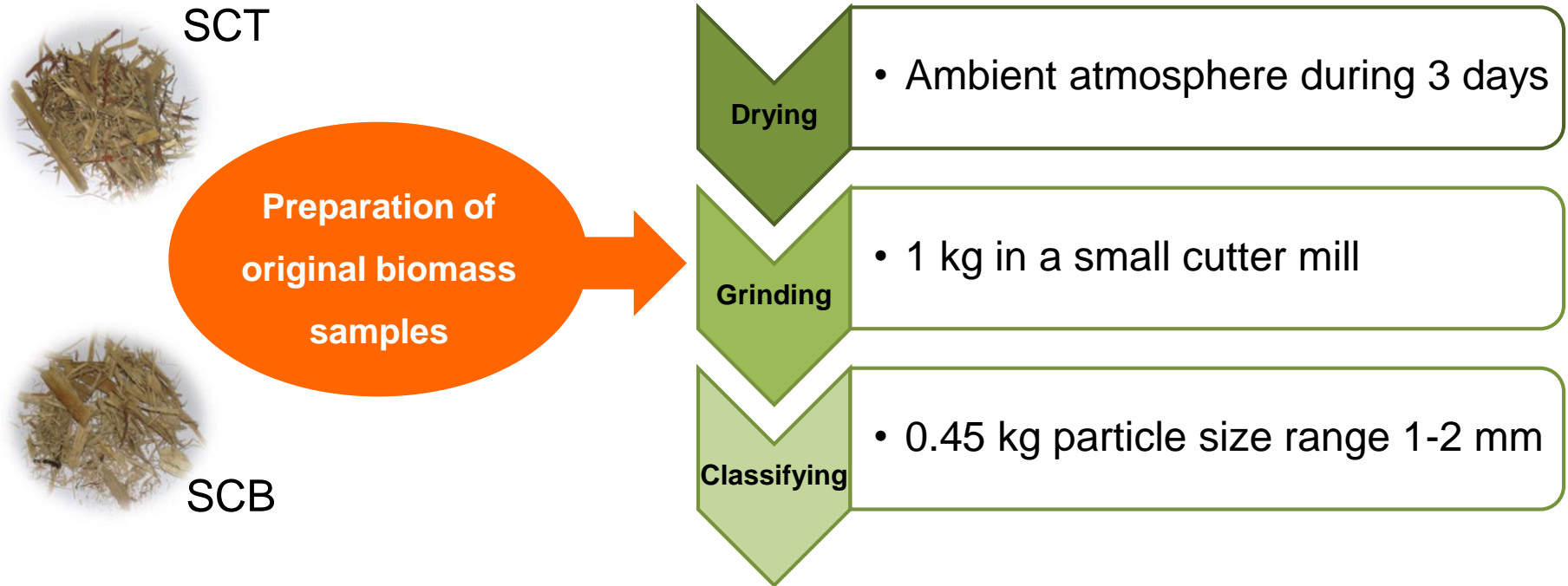


Eom et al. **2012**. Effect of essential inorganic metals on primary thermal degradation of lignocellulosic biomass. *Bioresource Technology*, 104(0), 687-694.

**General objective:** to evaluate the effect of demineralization of sugarcane bagasse and sugarcane trash on the chemical composition of the bio-oil, viz. by applying micro-pyrolysis (py-GC/MS).



# Feedstock preparation prior to leaching



*SCT: sugarcane trash*

*SCB: sugarcane bagasse*

## Characterization of untreated samples

Proximate analysis, wt. %			Ultimate analysis, wt. % d.b.			ICP-OES analysis, mg/kg		
	SCB	SCT		SCB	SCT		SCB	SCT
Volatile matter <sup>d.b.</sup>	79.8	73.9	C	44.1	40.1	K	1,800	2,300
Fixed carbon <sup>a</sup>	17.1	19.2	H	6.0	5.3	Na	32.1	56.8
Moisture <sup>a.r.</sup>	6.7	7.1	O <sup>a</sup>	47.7	48.8	Mg	287	492
Ash <sup>d.b.</sup>	<b>1.8</b>	<b>5.3</b>	N	0.2	0.5	Fe	327	245
						Al	279	216
						Si	8,600	13,400

**Ash of pine wood: 0.33 wt.% d.b.**

<sup>d.b.</sup> On dry basis

<sup>a</sup> Calculated by difference

<sup>a.r.</sup> As received basis

Sulphur: below detection limit



# Leaching conditions

SCT



Water (W)

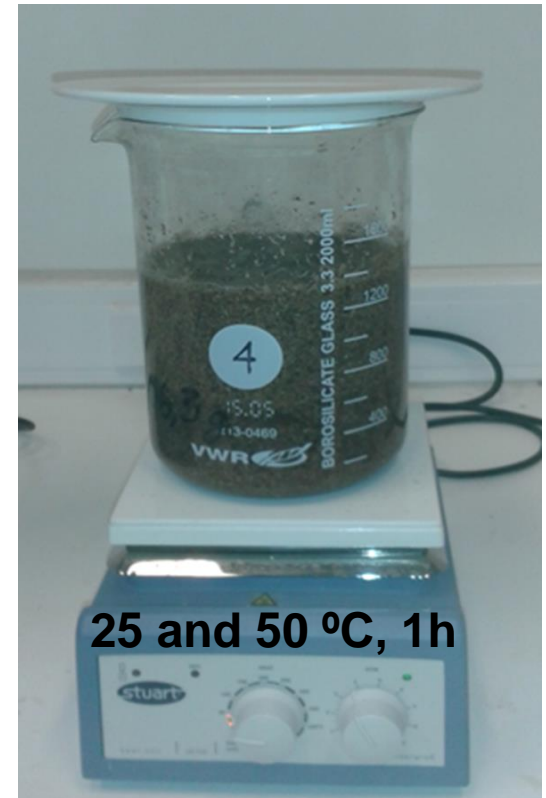
HCl, (5M)

H<sub>2</sub>SO<sub>4</sub>, (5M)

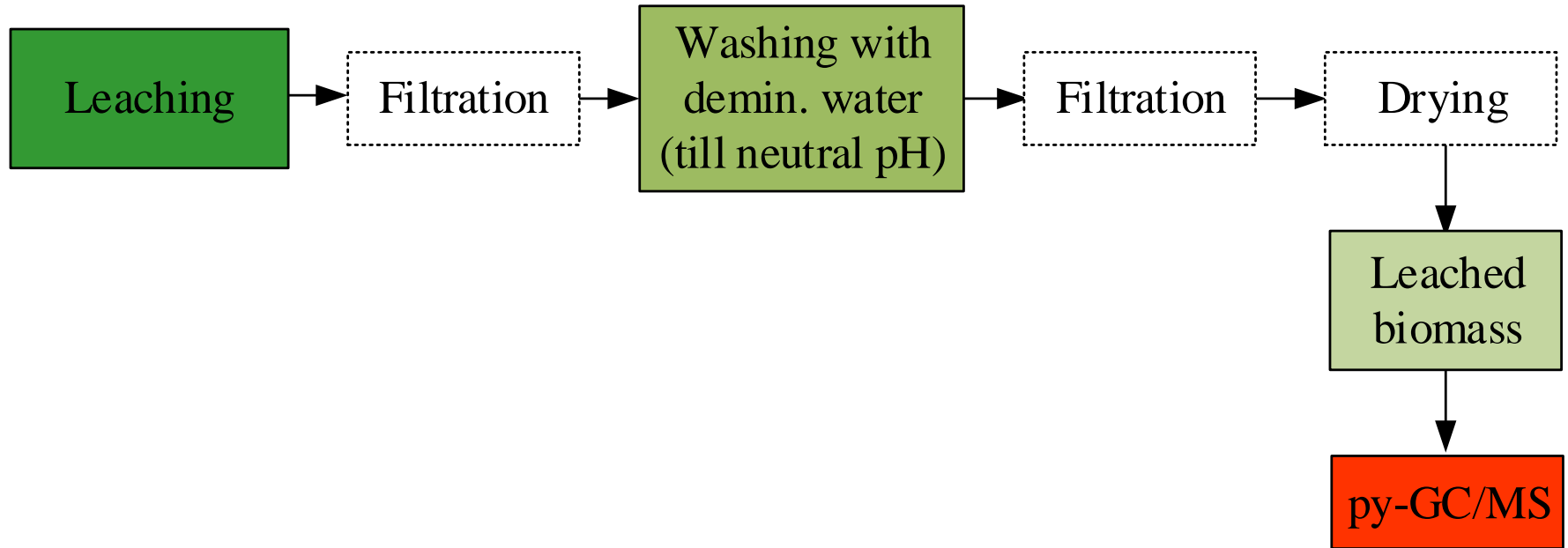
Citric acid (CA), (1M)



SCB



# Biomass processing



# Proximate (wt.% ) and ICP-OES analysis (mg/kg) of biomass leached at 25 °C, 1h

Washing agent	SCB					SCT				
	Unt.	Water	HCl	CA	H <sub>2</sub> SO <sub>4</sub>	Unt.	Water	HCl	CA	H <sub>2</sub> SO <sub>4</sub>
V.M. <sup>d.b.</sup>	<b>79.8</b>	82.3	80.7	80.7	79.8	<b>73.9</b>	77.4	76.1	76.9	76.7
F.C. <sup>a</sup>	<b>17.1</b>	17.1	17.6	17.6	18.6	<b>19.2</b>	17.8	19.4	18.4	19.1
Moisture <sup>a.r.</sup>	<b>6.7</b>	3.1	3.0	3.4	2.8	<b>7.1</b>	3.4	3.9	5.0	3.2
Ash <sup>d.b.</sup>	<b>1.8</b>	1.3	1.1	1.1	1.1	<b>5.3</b>	4.2	3.6	3.7	3.6
A.R. (%)	-	33.6	48.4	48.4	48.4	-	29.4	47.4	38.9	45.6
K	<b>1,800</b>	293.0	17.5	17.0	15.7	<b>2,300</b>	142.0	16.2	17.6	13.2
Na	<b>32.1</b>	15.8	18.6	<5.0	8.02	<b>56.8</b>	6.14	49.2	< 5.0	24.3
Mg	<b>287</b>	198.0	25.1	68.1	26.4	<b>492</b>	330.0	14.1	29.6	19.9
Fe	<b>327</b>	209.0	110.0	123.0	101.0	<b>245</b>	122.0	61.5	104.6	73.3
Al	<b>279</b>	180.0	110.0	131.0	122.0	<b>216</b>	101.0	54.0	78.0	59.8
Si	<b>8,600</b>	4,500	3,600	4,400	4,400	<b>13,400</b>	15,600	14,600	6,300	11,000

V.M. Volatile matter

F.C. Fixed carbón

A.R. Ash removal

Unt. Untreated

<sup>d.b.</sup> On dry basis

<sup>a</sup> Calculated by difference

<sup>a.r.</sup> As received basis

Sulphur: below detection limit

# Proximate (wt.% ) and ICP-OES analysis (mg/kg) of biomass leached at 50 °C, 1h

Washing agent	SCB					SCT				
	Unt.	Water	HCl	CA	H <sub>2</sub> SO <sub>4</sub>	Unt.	Water	HCl	CA	H <sub>2</sub> SO <sub>4</sub>
V.M. <sup>d.b.</sup>	<b>79.8</b>	81.6	80.0	80.5	73.9	<b>73.9</b>	76.8	74.3	75.4	70.9
F.C. <sup>a</sup>	<b>17.1</b>	16.3	18.1	17.7	24.1	<b>19.2</b>	18.3	20.1	18.7	23.2
Moisture <sup>a.r.</sup>	<b>6.7</b>	4.3	2.4	3.8	2.0	<b>7.1</b>	2.9	3.1	5.2	3.1
Ash <sup>d.b.</sup>	<b>1.8</b>	1.4	1.4	1.0	2.4	<b>5.3</b>	4.4	5.1	3.8	5.2
A.R. (%)	-	31.1	48.2	50.1	54.7	-	26.9	45.2	39.0	46.6
K	<b>1,800</b>	125.0	26.9	11.2	11.8	<b>2,300</b>	219.0	16.7	12.3	15.2
Na	<b>32.1</b>	16.0	21.5	4.8	5.3	<b>56.8</b>	<b>15.1</b>	46.7	< 5.0	<5.0
Mg	<b>287</b>	143.0	18.8	<b>25.2</b>	<b>15.3</b>	<b>492</b>	256.0	9.4	56.9	28.4
Fe	<b>327</b>	138.0	74.5	68.7	52.5	<b>245</b>	113.0	73.7	80.0	89.9
Al	<b>279</b>	119.0	138.0	<b>65.0</b>	<b>103.3</b>	<b>216</b>	<b>206.6</b>	62.7	55.1	<b>158.3</b>
Si	<b>8,600</b>	3,275	3,700	2,916	3,956	<b>13,400</b>	15,600	13,700	<b>&lt;200.0</b>	19,110

V.M. Volatile matter

F.C. Fixed carbón

A.R. Ash removal

Unt. Untreated

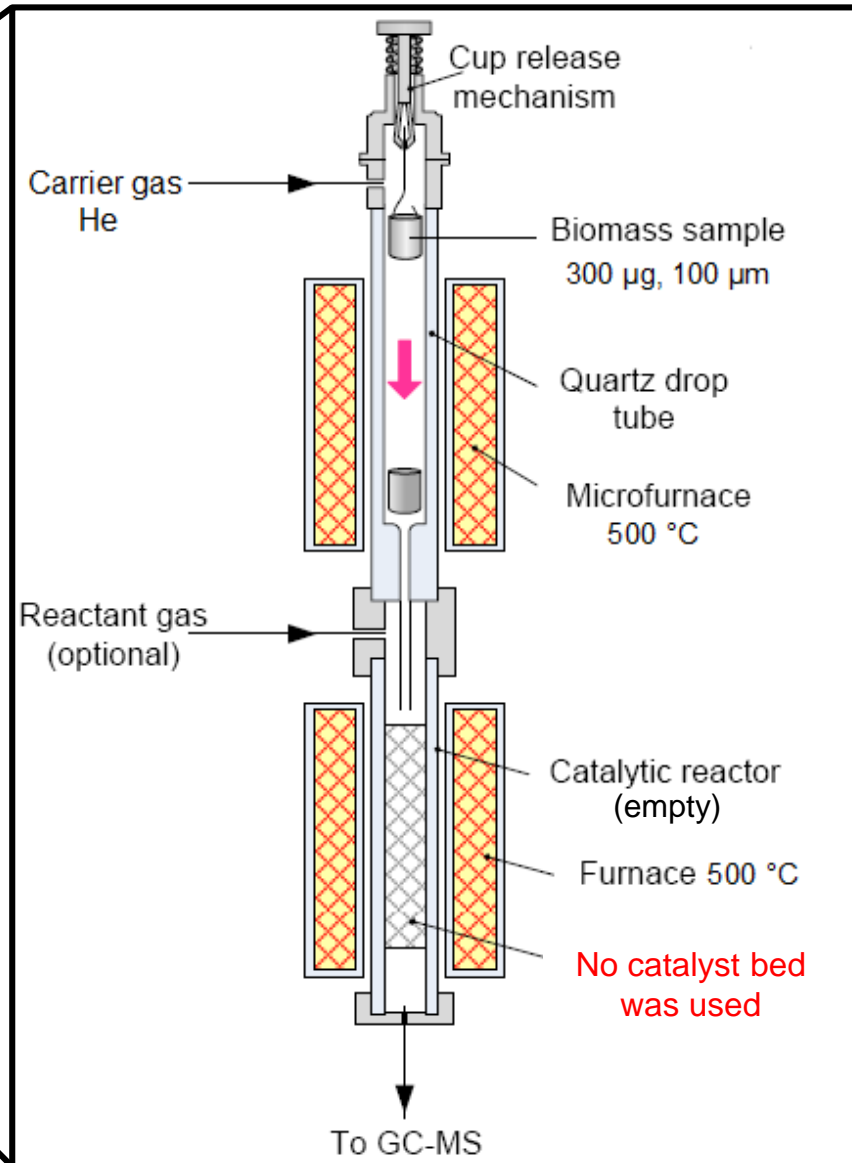
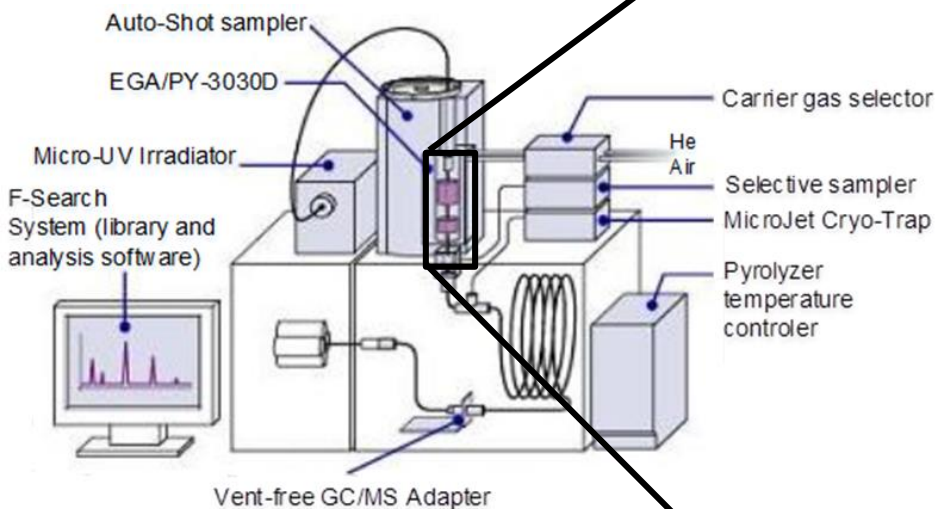
<sup>d.b.</sup> On dry basis

<sup>a</sup> Calculated by difference

<sup>a.r.</sup> As received basis

Sulphur: below detection limit

# Micro-pyrolysis setup



Frontier Laboratories. Multi-Shot Pyrolyzer, EGA/PY-3030D.  
[www.frontier-lab.com](http://www.frontier-lab.com), 2016

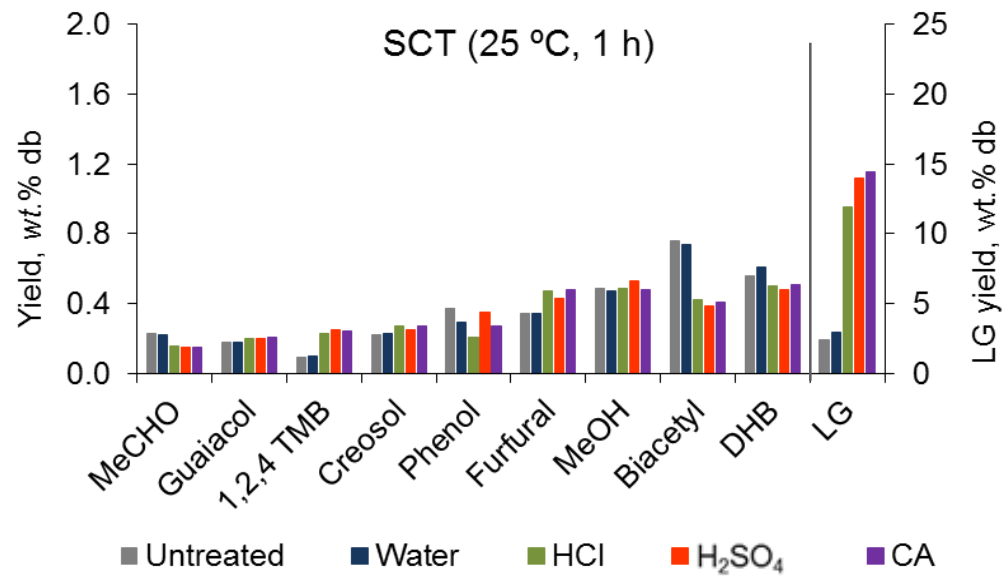
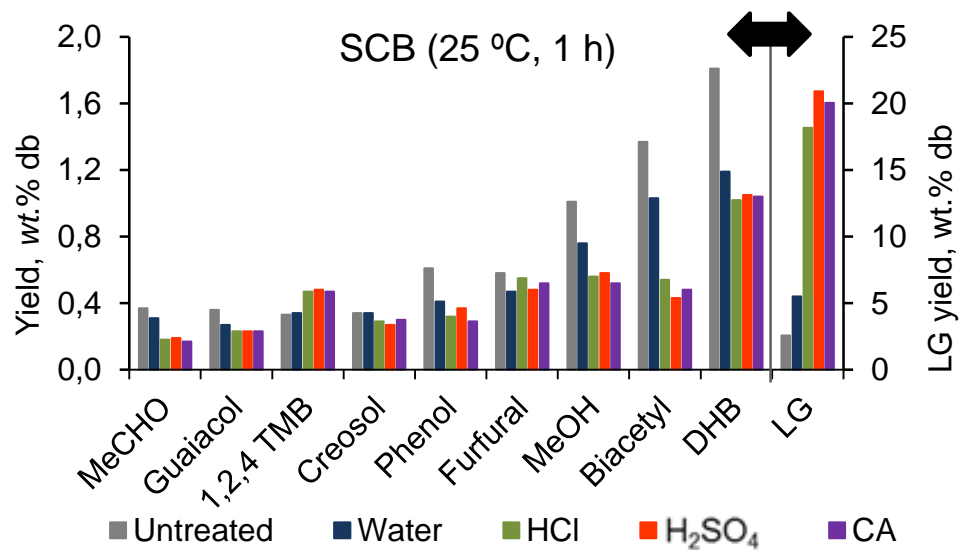


## Calibrated vapour phase compounds

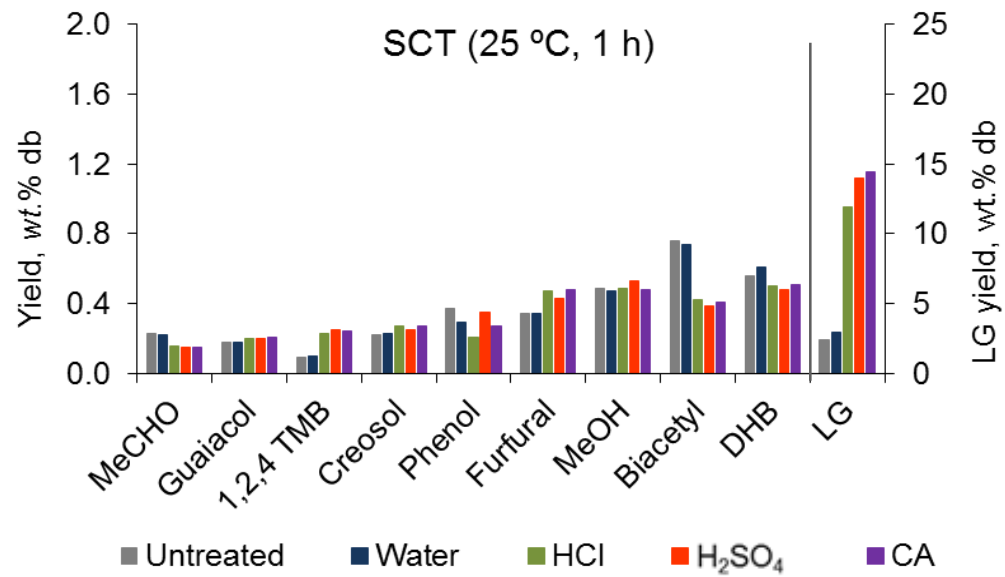
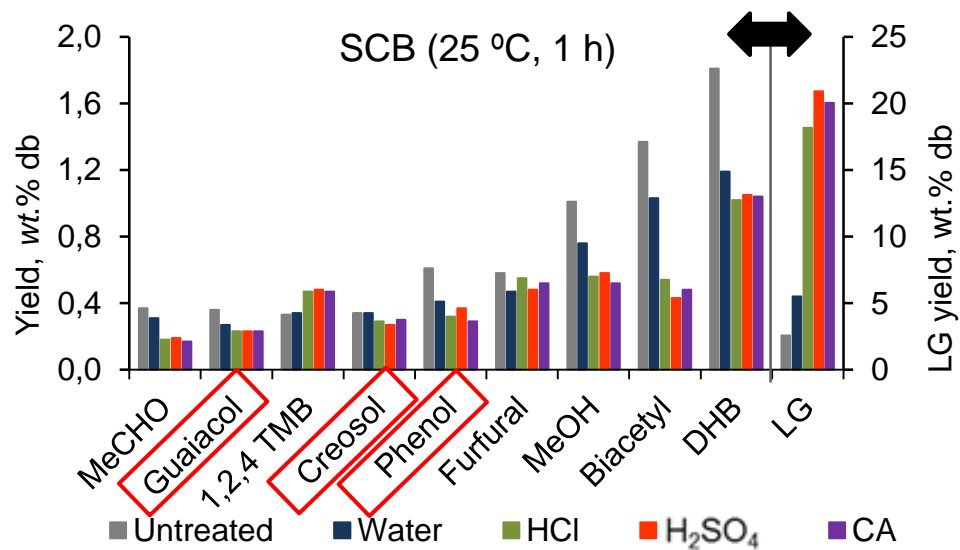
Compound name (Trivial name or abbrev.)	Formula	Origin <sup>a</sup>
Acetaldehyde (MeCHO)	C <sub>2</sub> H <sub>4</sub> O	C
Methanol (MeOH)	CH <sub>4</sub> O	C/L
2,3-Butanedione (Biacetyl)	C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	C
2-Furaldehyde (Furfural)	C <sub>5</sub> H <sub>4</sub> O <sub>2</sub>	C
Phenol	C <sub>6</sub> H <sub>6</sub> O	L-H
2-Methoxyphenol (Guaiacol)	C <sub>7</sub> H <sub>8</sub> O <sub>2</sub>	L-G
2-Methoxy-4-methylphenol (Creosol)	C <sub>8</sub> H <sub>10</sub> O <sub>2</sub>	L-G
2,3-dihydrobenzofuran (DHB)	C <sub>8</sub> H <sub>8</sub> O	Un.
1,2,4-trimethoxybenzene (1,2,4TMB)	C <sub>9</sub> H <sub>12</sub> O <sub>3</sub>	Un.
1,6-anhydro-β-D-Glucopyranose, Levoglucosan (LG)	C <sub>6</sub> H <sub>10</sub> O <sub>5</sub>	C

<sup>a</sup> C, L, L-H, L-G, Un. represent cellulose, lignin, p-hydroxyphenyl lignin, guaiacyl lignin and unknown, respectively.

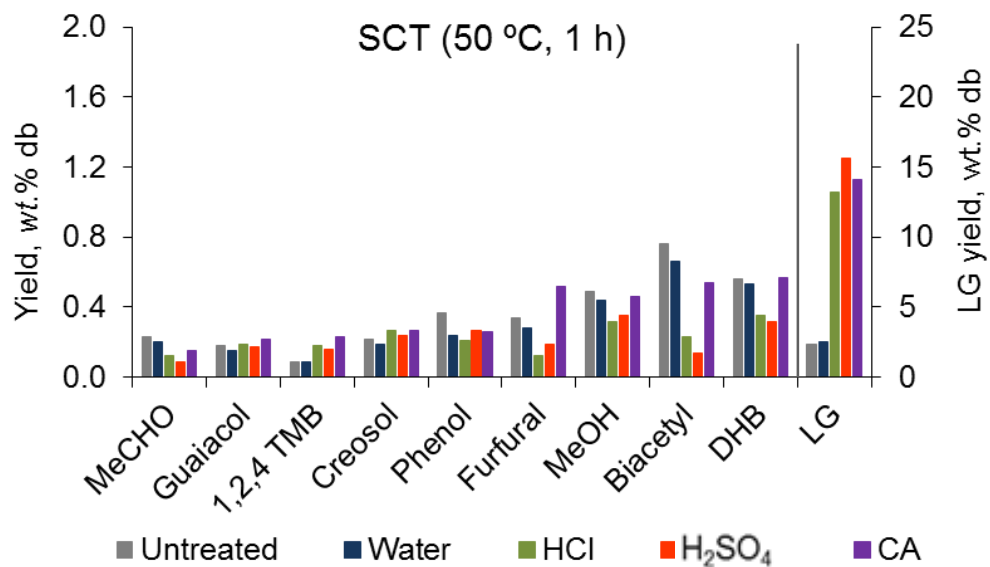
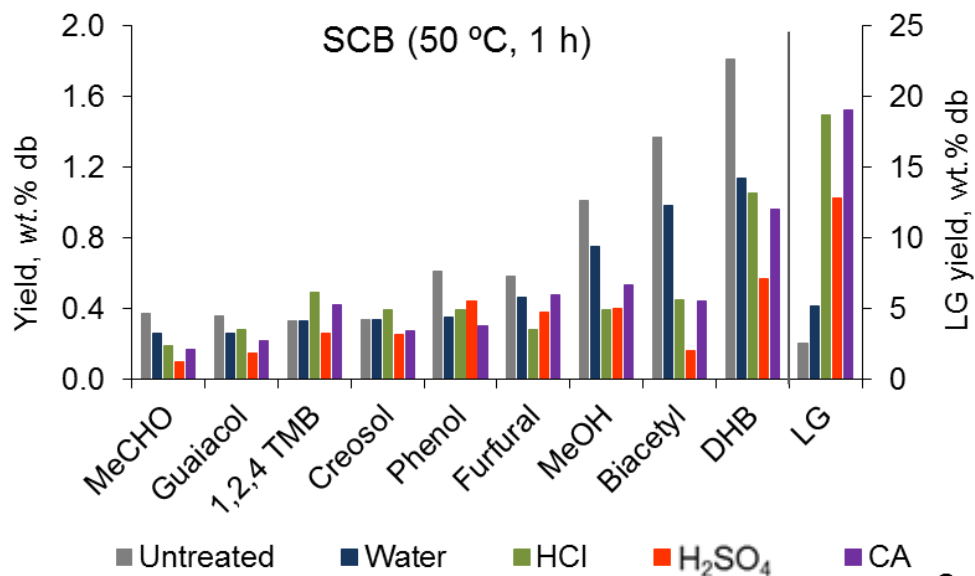
# Yield of calibrated vapour phase compounds of leached biomass (wt.% on d.b.)



# Yield of calibrated vapour phase compounds of leached biomass (wt.% on d.b.)

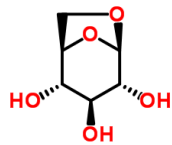


# Yield of calibrated vapour phase compounds of leached biomass (wt.% on d.b.)



# Effect of acid leaching

Levogluconan



- Decreasing in AAEM = decreased catalysis

- Decreasing of cellulose crystallinity ?

- Prehydrolysis ?



# Conclusions

- The results of this work suggest that chemical treatment of SCB and SCT, either with inorganic (HCl, H<sub>2</sub>SO<sub>4</sub>) or organic (CA) acids, increases the yields of levoglucosan in their volatiles between 10-18 (SCB) and 9-13 (SCT) *wt.%*.
- SCT also is a valuable feedstock for fast pyrolysis as its spectrum of pyrolysis vapour compounds is similar to that of SCB.
- Citric acid could be considered as a leaching solution.
- The volatile compounds distribution of citric acid treated biomass showed small variations upon changing the pretreatment temperature.
- The yield of lignin-derived compounds decreases slightly when sugarcane biomass is treated.

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